



CHEMICAL SCIENCES

Determination of phytochemical contents by LC/QTOF/MS and evaluation of *in-vitro* biological activities of 2 *Peltigera* lichens from Bursa

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Abstract: Lichens are symbiotic associations of algae and fungi. They are edible as food and have been used in traditional medicine for years. It is aimed to screen *Peltigera praetextata* (Flörke ex Sommerf.) Zopfand and *Peltigera elisabethae* Gyeln. phytochemically by LC/QTOF/MS and according to the constituents to evaluate the antioxidant, tyrosinase inhibitory, and antibacterial activities. In total 54 of metabolites detected by LC/QTOF/MS were common in both species. According to LC/QTOF/MS scanning results, alkaloids, iridoid glycosides, phenolics, cyanogenetic glycosides, and terpenic structures were detected. DPPH, ABTS, superoxide radical scavenging activities, and metal chelating capacity IC_{50} values were 84.55, 9.349; 51.27, 9.127; 95.01, 58.65 and 20.57, 70.08 $\mu\text{g}/\text{mL}$, respectively. The CUPRAC reducing power was determined as 4.69 and 9.57 $\text{TEAC}_{\text{CUPRAC}}$, respectively. Tyrosinase inhibitor activity were found to be 86.95 and 196.7 $\mu\text{g}/\text{mL}$. Both lichens did not show antimicrobial effects. As a result of the antioxidant and tyrosinase inhibitor activities it was seen that their activities were significant and further *in vivo* studies could be carried out on this lichens.

Key words: Antioxidant, LC/QTOF/MS, Lichen phenolics, *Peltigera elisabethae* Gyeln., *Peltigera praetextata* (Flörke ex Sommerf.), Tyrosinase inhibitor.

INTRODUCTION

Lichens are symbiotic associations that spread almost worldwide with twenty-five thousand taxa. Lichens are formed due to the physiological and morphological associations of algae and fungi. They can occur as long as there is sufficient moisture. With the lichen acids they secrete and they can even decompose the stones and form humus soil. Therefore, they are of great importance as pioneer organisms. According to the morphological forms of the thallus; they are divided into crustose, foliose, and fruticose lichens. According to their anatomical structures, they are divided into the homeomeric type and heteromeric type (Ivanova & Ivanov 2009, Emsen 2019). Lichens

also take place in ethnobotanical uses. Because it is rich in calcium and iron, it is consumed as food after boiling. *Peltigera*, like other lichens, is a genus often used in traditional medicine. It is known that they are used especially for skin disorders, wound healing, gynecological diseases, respiratory tract and digestive regulators (Rankovic & Kosanic 2015). They are very rich in secondary metabolites. They consist of aliphatic and aromatic compounds. Depsids, tridepsids and depsidones, dibenzofuran, cromon, and quinone derivatives are lichen secondary metabolites (Huneck & Yoshimura 1996). Lichen secondary metabolites have antioxidant, anticancer, antibiotic, antimicrobial, enzyme inhibitor, antiviral, and anti-genotoxic

effects (Huneck & Yoshimura 1996, Rankovic & Kosanic 2015, Sivas 2019).

In nature, most lichens are edible and provide nutrition for microflora and mammals. It is known that it is often consumed as a human food item. *P. rufescens* and *P. apthosa* lichens are used as flour. In addition to being consumed as food (flour etc.), their use in traditional medicine is also known. For example, it is known that *P. canina* is used in treating rabies among people, and it is known that it got its name from there. In addition, traditionally, *Peltigera* type lichens have been shown to have anti-inflammatory and antibacterial uses in jaundice (Rankovic & Kosanic 2015, Sivas 2019).

Today, lichens are one of the new focuses of attention, and the lack of studies about them has attracted our attention to contribute to the literature.

In this study, phytochemical screening of *P. praetextata* (PP) and *P. elisabethae* (PE) was performed with LC/QTOF/MS, and they were investigated for their antioxidant, antimicrobial, and tyrosinase inhibitor activities.

List of abbreviations

ABTS: 2,2'-Azino-bis(3-ethylbenzothiazoline-6-sulfonic acid) diammonium salt
CUPRAC: Copper(II) Ion Reducing Capacity

DPPH: 2,2'-diphenyl-1-picrylhydrazil

HPLC: High Performance Liquid Chromatography
NBT; Nitroblue tetrazolium

PE: *P. Elisabetheae*

PP: *P. praetextata*

TFC: Total flavonoid content
TPC: Total phenolic content

Chemicals and reagents

Methanol (Sigma), Ethanol (Sigma), LC grade Formic acid (Sigma), LC grade acetonitrile (Sigma), Folin Ciocalteu reagent (Merck), Na₂CO₃ (Merck), Gallic acid (Sigma), AlCl₃ (Merck),

Quercetin (Sigma), DPPH (Sigma), DMSO (Sigma), ABTS(Sigma), Neocuproine(Sigma), Copper (II) chloride(Sigma), EDTA(Sigma), NBT (Nitroblue tetrazolium) (Sigma), Riboflavin(Sigma), Phosphate buffer(Sigma), Ascorbic acid(Sigma), Ferrozine(Sigma), L-Dopa(Sigma), Kojic acid (Sigma), tyrosinase enzyme(Sigma).

MATERIALS AND METHODS

Lichen collection, determination, drying, and extraction

PP and PE species distributed in Bursa province were investigated. The collected lichen samples were identified with the help of identification keys in various flora books (Wirth 1995, Brodo et al. 2001, Smith et al. 2009).

Prof. Dr. Şule Öztürk and Assoc. Prof. Dr. Seyhan Oran identified them. The collected samples, especially the very moist ones of the species that spread on the soil, were kept in laboratory conditions for four days and dried as a pre-treatment procedure. The thalli of the collected lichen samples were carefully cleaned from foreign materials such as moss, soil, and tree bark in a Leica MZ 6 stereomicroscope with dissection forceps, placed in paper bags, and stored in the refrigerator until the laboratory experiments began.

Lichen extraction

For the extraction process, 1 gram of powdered lichen thallus was extracted three times with methanol in an ultrasonic bath (Bandolin Sonorex) (25 °C) for 30 minutes. After each filtering, the solvent was removed in the rotavapor (Buchi) at 35-40 °C, and the collected extracts were combined and then stored at + 4 °C (Adams et al. 1993, Zagoskina et al. 2013).

Qualification of compounds by LC/QTOF/MS

Chromatographic separation was carried out using an HPLC Agilent 1260 Infinity series (Agilent Technologies, Santa Clara, CA, USA) equipped with a binary pump, and an online degasser, an autosampler, and a Poroshell 120 EC-C18 (3.0X100 mm, particle size) 2.7 μm (Agilent Technologies). A mobile phase system composed of water with 0.1% formic acid (A) and acetonitrile (B) was used in a gradient elution mode as follows: 0–0.5 min, 5% B; 0.5–7 min, 25% B; 7–16 min, 50% B; 16–23 min, 75% B; 23–30 min, 95% B; 30–40 min, 5% B for equilibration of the column. The column temperature was maintained at 35 °C. The injection volume was 10 μL , and the flow rate was 0.4 mL/min.

Ionization of chromatographic eluates was performed using an Agilent 6550 iFunnel high-resolution Accurate-Mass QTOF-MS, equipped with an Agilent Dual Jet Stream, and electrospray ionization (Dual AJS ESI) interface operating in negative ion was used at the following conditions: drying gas flow, 14.0 L/min; nebulizer pressure, 35 psi; gas drying temperature, 290 °C; sheath gas temperature, 400 °C; sheath gas flow, nitrogen at 14 L/min and nozzle voltage 1000 V. All spectra were collected in targeted MS/MS mode from m/z 50 to 1800 scan range for inducing MS/MS data collection. The collision energy was 20 eV during analysis.

Integration and data evaluation were performed using MassHunter Workstation software, and the MassHunter METLIN and the Accurate Mass Personal Compound Database and Library (METLIN_AM_PCDL) were used to detect phenolic compounds. (Agilent Technologies, Santa Clara, CA, USA). Agilent Dual Jet Stream electrospray ionization (Dual AJS ESI) interface operating in negative and positive ions.

Determination of total phenol and flavonoid content

Determination of Total Phenolic Content (TPC)

The analysis was carried out according to the Folin-Ciocalteu method. 2.8 mL of deionized water was added to 100 μL of extract (concentration range of 5 $\mu\text{g}/\text{mL}$ –1mg/mL). Then, 2 mL of 2% Na_2CO_3 and 0.1 mL of 50% Folin reagent were added, mixed, and incubated at 25 °C for 30 minutes in the dark. Gallic acid calibration curve and total phenolic content of lichens were determined by measuring absorbance against water at 750 nm by Optima SP-3000 Nano Spectrophotometer (Cheung et al. 2003, Kanipandian et al. 2014). Results are expressed as mg gallic acid/g extract.

Determination of Total Flavonoid Content (TFC)

1.5 mL of 95% ethanol was added to 500 μL of extract solution (concentration range of 5 $\mu\text{g}/\text{mL}$ – 1mg/ml). After, 100 μL of AlCl_3 and 2.8 mL of deionized water were added to it and left to incubate for 40 minutes. The absorbance was measured against ethanol at 415 nm by Optima SP-3000 Nano Spectrophotometer. Quercetin was used as a standard (Cheung et al. 2003, Kanipandian et al. 2014). The results are reported as mg quercetin/g extract.

In vitro Antioxidant Activity Studies

DPPH (2,2'-diphenyl-1-picrylhydrazil) Radical Scavenging Activity

The method described by Esmaeli et al. was used with little modification (Esmaeili & Khadadadi 2002). DPPH solution prepared in DMSO was added to the tested samples in a certain concentration range (10 $\mu\text{g}/\text{mL}$ – 1 mg/mL) and then incubated in the dark for 30 minutes. The absorbance of the resulting mixture was then measured at 517 nm by Optima SP-3000 Nano

Spectrophotometer and methanol used as a blank. α -tocopherol was used as a standard. Results are given as IC_{50} ($\mu\text{g/mL}$) (GraphPad Prism 5).

ABTS⁺ Radical Cation Scavenging Activity

It was evaluated by modifying the ABTS⁺ decolorization method of Rer et al. (1999). In this method, ABTS radical was prepared by reacting stock ABTS solution with potassium persulfate (Rer et al. 1999). It was diluted with ethanol until an absorbance of 0.750 at 734 nm was obtained. Then, 0.1 mL of extract (concentration range of 20 $\mu\text{g/mL}$ -1 mg/mL) and 4 mL of ethanol were added to 1 mL of diluted ABTS solution, and its absorbance at 734 nm at 6 minutes was read by Optima SP-3000 Nano Spectrophotometer. α -tocopherol was used as a standard. Results are given as IC_{50} ($\mu\text{g/mL}$) (GraphPad Prism 5) Ethanol was used as a blank.

CUPRAC (Copper(II) Ion Reducing Capacity) Method

The copper(II) reducing power was determined according to the method of Apak et al. (2004). After the mixture of neocuprine and Copper (II) was prepared at pH 7, sample solution (5 $\mu\text{g/mL}$ -1 mg/mL) was added and incubated for 30 minutes. The absorbance against the blank methanol at 450 nm was measured by Optima SP-3000 Nano Spectrophotometer. Trolox solution was used as standard. Results are given as $TEAC_{CUPRAC}$.

Superoxide Radical Scavenging Activity

It was carried out according to the research conducted by Patel et al. (2010). 10 μL of extracts and standards at different concentrations, 15 μL 12 mM EDTA, 10 μL 0,1 mg/mL NBT (nitro blue tetrazolium), 5 μL 0.2 mg/mL riboflavin, and 160 μL 0.067mM Potassium Phosphate Buffer (pH 7.4) are placed in a Thermo Scientific- Varioscan

Flash microplate reader and incubated at fluorescence light for 5 minutes. Ascorbic Acid was used as a standard. The absorbance at 560 nm was then measured by Optima SP-3000 Nano Spectrophotometer. Results are given as IC_{50} ($\mu\text{g/mL}$) (GraphPad Prism 5) (Patel et al. 2010).

Metal Chelating Activity

The metal chelating activity was modified using the Fe (II)-Ferrozine method. In this method, 5 mL of test solutions at concentrations of 31.25-500 $\mu\text{g/mL}$ were added to 100 μL of 2 mM FeCl_2 and incubated for 30 minutes at room temperature in the dark. At the end of the period, 200 μL of 5 mM ferrozine solution was added and incubated again for 10 minutes in the dark and at room temperature. EDTA was used as a standard. Its absorbance at 562 nm was measured by Optima SP-3000 Nano Spectrophotometer. Results are given as IC_{50} ($\mu\text{g/mL}$) (GraphPad Prism 5) (Decker, 1997).

In vitro Tyrosinase Inhibitor Activity

The method of Masuda et al. (2005) was modified. 110 μL phosphate buffer (0.01 M, pH: 6,8), 10 μL plant extract (different concentrations), and 20 μL tyrosinase solution (200 unit/mL) were mixed. After 10 minutes of incubation at 37 °C, the reaction was initiated by adding 20 μL of L-Dopa and allowed to incubate at 37 °C for 10 minutes. The absorbance was measured at 475 nm by Thermo Scientific Varioskan Flash Microplate Reader. Kojic Acid was used as a standard. Results are given as IC_{50} ($\mu\text{g/mL}$) (GraphPad Prism 5) (Masuda et al. 2005).

Antimicrobial Activity

In antimicrobial tests, Gram-positive bacteria *Bacillus cereus* (ATCC 7064), *Enterococcus faecalis* (ATCC 29212), *Staphylococcus aureus* (ATCC 6538-P), Gram-negative bacteria *Escherichia coli* (ATCC 25922), *Klebsiella pneumoniae* (ATCC 13883),

Pseudomonas aeruginosa (ATCC 27853), and two yeast strains *Candida albicans* (ATCC 90028), *Candida parapsilosis* (ATCC 22019) were used.

Agar Well Diffusion Method

Fresh cultures of bacteria and standard yeast strains were adjusted to a 0.5 McFarland turbidity by suspending in physiological saline water and inoculated under aseptic conditions on cation-added Mueller Hinton Agar (MHA) and Sabouraud Dextrose Agar (SDA) plates respectively. 50 μ l of the stock solutions (4mg/ml) of the extracts prepared in deionized and sterilized water were filled into the opened wells on the agar plates. The final extract loads were 200 μ g in each well. Ciprofloxacin 5 μ g and fluconazole 25 μ g were used as positive controls. An evaluation was made by measuring the zone diameters seen at the end of 24-48 hours of incubation at 37 $^{\circ}$ C (Korkak et al., 2022).

Microdilution Method

50 μ L cation-added Mueller Hinton Broth (MHB) medium for yeasts, and Sabouraud Dextrose Broth (SDB) medium for yeasts were added to the wells in a sterile U-bottom microplate. Stock solutions of the extracts in water (4mg/mL) were distributed on the microplate with

a concentration range of 2-2048 μ g/mL. Microorganisms, whose fresh cultures prepared from the stock were suspended in physiological saline and adjusted to 0.5 McFarland turbidity, were diluted 1:100 with MHB and SDB media, then distributed in equal amounts to the wells. The first well without visible turbidity at the end of 24-48 hours of incubation was considered as minimum inhibitory concentration (MIC). Growth control and medium controls were also included. Ciprofloxacin and fluconazole were used as positive controls (Korkak et al., 2022).

Statistical analysis

All results are expressed as mean \pm SD. One Way ANOVA test at a level of $p < 0.05$ in IBM SPSS (ver. 25). Were performed (Table V).

RESULTS AND DISCUSSION

As a result of the extractions, PE and PP extracts were obtained with a yield of 18.11% and 16.31% respectively. According to LC-QTOF-MS library scanning, a total of 441 items were identified (Supplementary files-Tables), including both secondary and primary metabolites. All chromatograms are listed in Figure 1, 2, 3 and 4. Alkaloids, iridoid glycosides, phenolics,

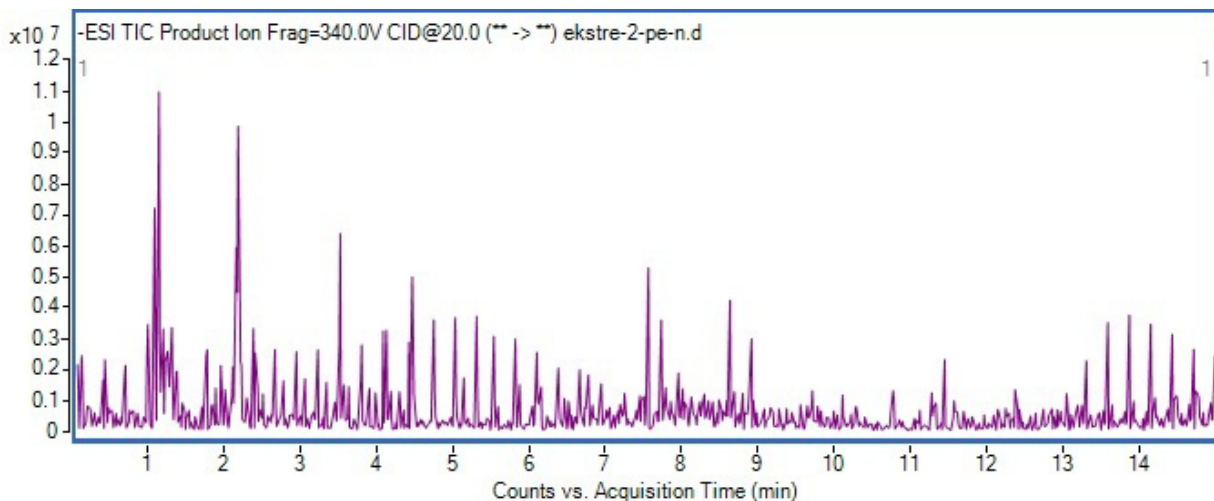


Figure 1. Negative ion mode LC/QTOF/MS Chromatogram of PE.

cyanogenetic glycosides, and terpenic compounds were determined as secondary metabolites. Xanthosine, C16 Sphinganine, Tortuosamine, Erythronolide B, Prosopinine, 14,19- Dihydrospidospermatine, Tetradecylamine, Fenpiclonil, Eudesmin, n-Pentadecylamine, Triclofos, Gemfibrozil M1, 9-nonadecene, 1-Eicosene, Spermine derivates, Diisobutyl/diisononyl phthalate, Luteolin derivate, Nonoxynol-9 and Harderoporphyrin were biologically active compounds and detected in both PP and PE (Figure 1,2,3,4). According to the literature research (Table I), it was seen that

these substances have antioxidant, tyrosinase inhibitor, and antimicrobial effects, and it was decided to screen the activities of the extracts based on this. TPC (Total phenolic content), TFC (Total flavonoid content) and *in vitro* antioxidant activities were evaluated and summarized (Tables II and III). It has been observed that the radical scavenging activities of PE, which has more TPC and TFC and contains more secondary metabolites known to be antioxidants, are much higher. Tyrosinase inhibitory activity results are summarized in Table IV. It was observed that activity of PP is higher than PE. This suggests

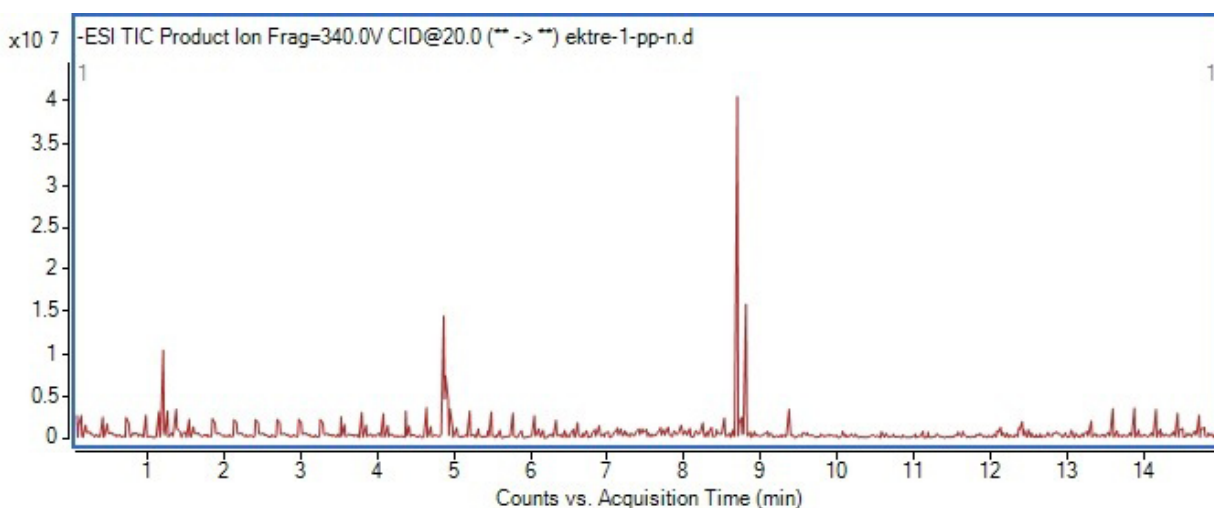


Figure 2. Negative ion mode LC/QTOF/MS Chromatogram of PP.

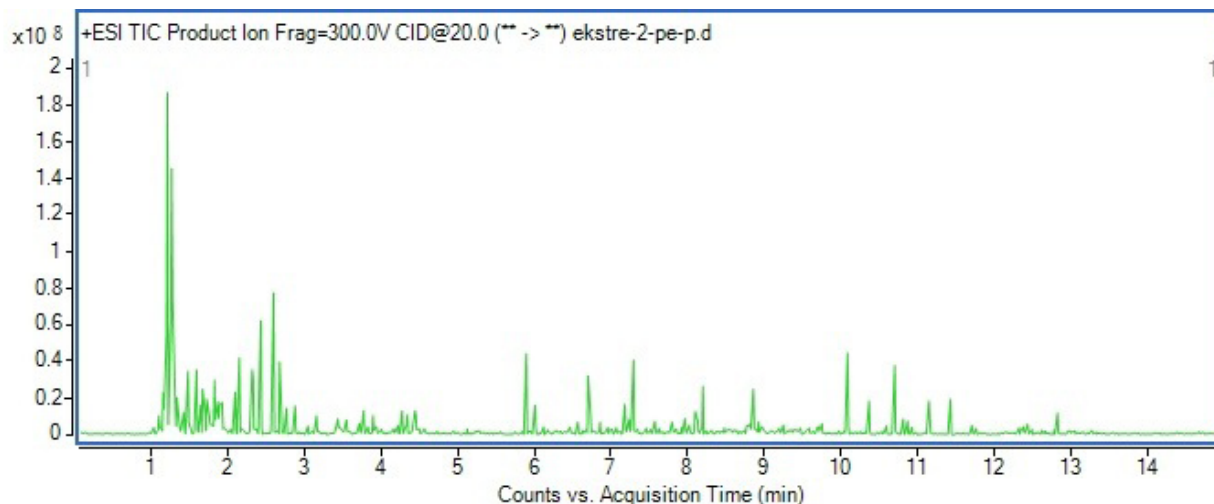


Figure 3. Positive ion mode LC/QTOF/MS Chromatogram of PE.

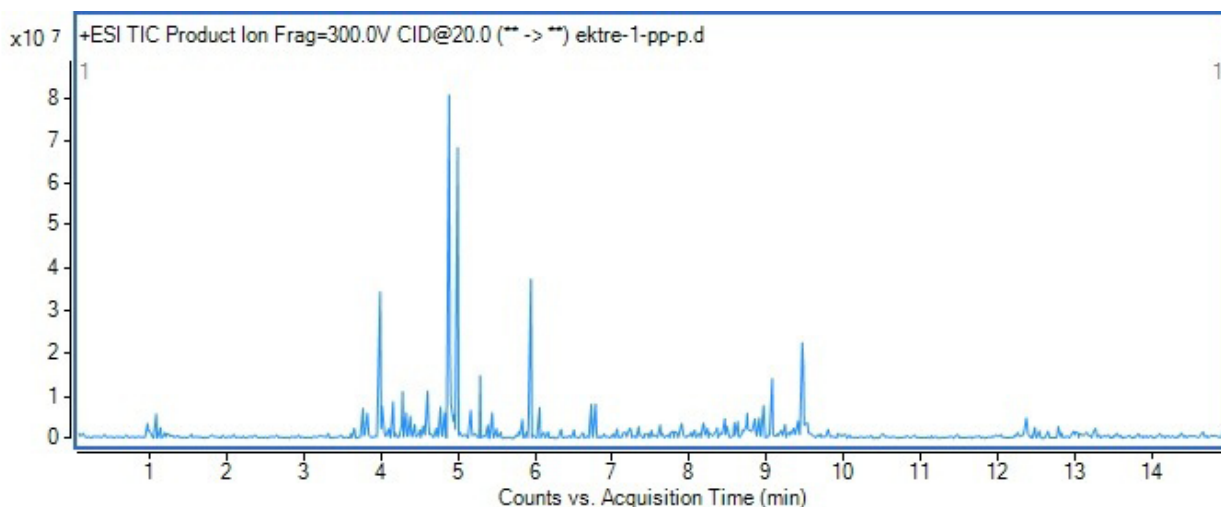


Figure 4. Positive ion mode LC/QTOF/MS Chromatogram of PP.

that this may be due to the amount of tyrosinase inhibitor substances and synergistic effects in their content. There are a few studies on PE and PP. When they were evaluated, it was determined that PP had an antimutagenic effect (Nardemir et al. 2015). Its antimutagenicity is important as it supports the consumability of edible lichens. In another study involving both PP and PE, their antiproliferative effects were determined. In the same study it was also stated that PP does not contain secondary metabolites in contrast to our study (Munzi et al. 2014).

When the studies on the other species in *Peltigera* genus were examined, it was seen that acetone extract of *P. horizontalis* were effective as an antioxidant. Still, it was not found to have an antibacterial effect in 5 different strains. In this study, while antioxidant activity was observed in our species, antibacterial activity was not observed in similar strains. The DPPH scavenging activity of *P. laciniata* ethanol extract was given as percent inhibition (80%). It can be seen that it shows a very strong scavenging

activity. Since our results are given as IC_{50} value in this study it cannot be compared exactly (Plaza et al. 2014). However we can indicate that the results are compatible. In a different study, DPPH radical scavenging and metal chelating activities of *P. canina* methanol extract were determined and expressed as IC_{50} value. These values were 42,37, and 50,33 mg/L, respectively (Emsen, 2019). Compared to our current research results, *P. canina*'s activity is between PP (84,55-20,57 $\mu\text{g}/\text{mL}$) and PE (9,349-70,08 $\mu\text{g}/\text{mL}$) in our research. A research with *P. rufescens* stated that it is effective as an antioxidant (Aydin & Turkez 2011). Compared to our results, it is true that both of them have antioxidant effects. As a result of a study with Nepal mountain lichens, *Peltigera* sp. was ineffective against *Bacillus subtilis* and *Staphylococcus aureus* as our study. Again in the same study, the radical scavenging effects of DPPH and ABTS were examined, and their IC_{50} values were reported as 5.6 and 6.9 $\mu\text{g}/\text{mL}$, respectively, closer to PE in our study (Paudel et al. 2012).

Table I. Compounds detected in PP and PE and biological activities in literature

	Compound name	Chemical structure	Detected Lichens	Retention time	Mass	Ion mode	Biological activity	Ref.for activity
1	Feruloylagmatine	Cinnamamides	PE	1.149	306.1702	Positive	Antioxidant and Antibacterial	Wang et al. 2020, Zeiss et al. 2021
2	Neolinustatin	cyanogenic glycoside	PE	1.372	423.1744	Positive	Antioxidant	Li et al. 2012a
3	Chavicol glycoside	Phenyl propanoid	PE	1.383	296.1267	Positive	Antioxidant and tyrosinase inhibitor	Oliveira et al 2021, Avetisyan et al 2017
4	Dehydrofalcarinone	ynone	PE	1.697	240.1515	Positive	Antioxidant	Khalil et al. 2022
5	Tetramethylquercetin 3-rutinoside	Flavonoid glycoside	PE	2.223	666.2150	Negative	Antioxidant	Olgun et al. 2018, Si et al. 2012
6	Atraric Acid	Phenolic	PE	2.725	196.0743	Positive	Antioxidant and tyrosinase inhibitor	Mitrovic et al. 2014, Lie et al. 2022
7	Levosimendan	Pyridazinone	PE	3.173	280.1068	Positive	Antioxidant	Abeer et al. 2017
8	Pyridyl nicotine and derivates	Alkoloid	PE	3.218	239.1419	Positive	Antioxidant and tyrosinase inhibitor	Lin et al. 2012
9	Donepezil	Piperidine derivate	PE	3.274	379.2130	Positive	Antioxidant	Umukoro et al. 2014
10	Nitrendipine	Pyridine derivate	PE	3.441	360.1328	Positive	Antioxidant	Unuvar et al. 2021
11	Austin	Terpene	PE	3.508	500.2031	Positive	Antioxidant and tyrosinase inhibitor	Burgos & Serranilos 2012, Kootheat et al. 2023
12	Anabsin	Sesquiterpenoid	PE	3.777	512.2750	Positive	Antioxidant	Brunet et al. 2005
13	Avenanthramide	Phenolic alkoloid	PE	3.788	361.1167	Positive	Antioxidant and tyrosinase inhibitor	Dimberg et al. 1993, Park et al 2021
14	Tocopheronic Acid	Sesquiterpenoid	PE	3.889	294.1471	Positive	Antioxidant	Al-Fatlawi & Al-Fatlawi 2020
15	3- methoxy chalcones	Chalcone	PE	3.934	448.1729	Positive	Antioxidant	Bandgar et al. 2010, Akhtar et al. 2015
16	Jasmonic acid glucoside	Jasmonate	PP	4.034	388.1729	Negative	Antioxidant	Sirhindi et al. 2016
17	Hordatine B	heterodimer of feruloylagmatine	PE	4.213	580.3141	Positive	Antioxidant	Wang et al. 2020
18	5,6,7-Trimethoxycoumarin	Coumarin	PE	4.437	236.0692	Positive	Antioxidant and tyrosinase inhibitor	Matos et al. 2017, Menezes & Diederich 2019
19	Mycophenolate mofetil	Carboxylic ester	PE	4.549	433.2096	Positive	Antioxidant	Dalmarco et al. 2009
20	Marmin	Coumarin	PP	4.779	360.1561	Positive	Antioxidant	Chian et al. 2018
21	Xanthosine	purine nucleoside	PP/PE	5.048	284.0753	Negative	Antibrowning agent	Tao et al. 2021
22	Cyproterone acetate	Steroidal	PE	5.242	416.1751	Positive	Tyrosinase inhibitor	Tadokoro et al. 2003
23	Hexyl glucoside	alkyl polyglycosides	PP	5.622	264.1584	Positive	Antioxidant	Adem et al. 2014
24	Apigenin	Flavonoid	PE	5.857	270.0539	Positive	Antioxidant and tyrosinase inhibitor	Sadasivam & Kumeresan 2011, Karaoglan &Koca 2020
25	Dihydroartemisinin	Sesquiterpenoid	PE	5.969	284.1629	Positive	Antioxidant	Ittarat et al. 2003
26	C16 Sphinganine	Amino alcohol	PP/PE	6.115	273.2672	Positive	Antioxidant and antibacterial	Tamsir et al. 2020, Barreto-Bergter et al. 2006
27	Cappariloside A	Alkoloid	PE	6.193	334.1170	Positive	Antioxidant and tyrosinase inhibitor	Aliyazicioglu et al 2013, Kirkan et al. 2021
28	Tortuosamine	Alkoloid	PP/PE	6.233	326.1978	Negative	Antioxidant and Antimalarial	Olatunji et al. 2021
29	Erythronolide B	Macrolide	PP/PE	6.278	402.2605	Negative	Antibacterial effect inducer	Saito & Mitsuhashi 1971
30	Prosopinine	Alkoloid	PP/PE	6.345	287.2465	Positive	Tyrosinase inhibitor	Saleem et al. 2021

Table I. Continuation.

31	14,19-Dihydroaspidospermatine	Alkoid	PP/PE	6.402	340.2136	Negative	Antibacterial	Tanaka et al. 2006
32	Tetradecylamine	Alkyl amine	PP/PE	6.450	213.2464	Positive	Antibacterial	Borick et al. 1959
33	Perilloside C	Terpene glycoside	PP	6.457	316.1893	Positive	Antioxidant	Hou et al. 2022
34	Fenpiclonil	Pyrrole	PP/PE	6.718	235.9907	Negative	Antibacterial	Leroux et al. 1992
35	Eudesmin	Lignan	PP/PE	6.726	386.1725	Positive	Tyrosinase inhibitor, Antibacterial	Li et al. 2013, Patel & Patel 2022
36	Oleamide	Fatty acid amide	PE	6.808	281.2719	Positive	Antioxidant	Reyes-Soto et al. 2022
37	Vialinin A	Terpene	PE	6.909	562.1624	Positive	Antioxidant	Sonowal et al. 2018
38	n-Pentadecylamine	Alkyl amine	PP/PE	7.020	227.2619	Positive	Antibacterial	Rammali et al. 2012
39	Sulfaphenazole	Sulfonamide	PP	7.064	314.0848	Negative	Antioxidant	Khan et al. 2007
40	Triclofos	Monoalkyl phosphate	PP/PE	7.134	227.8907	Negative	Antibacterial	Leroux et al. 1992
41	Gemfibrozil M1	Aromatic ether	PP/PE	7.743	266.1522	Positive	Antioxidant	Ashdate et al. 2002
42	Dodecyl Phosphocholine	Ethanol amine	PE	7.747	352.2614	Positive	Antioxidant	Balakrishna et al. 2017
43	9-nonadecene	Fatty acid	PP/PE	8.027	266.2979	Positive	Antioxidant and Antibacterial	Premathilaka & Silva 2016, Mahamuni 2015
44	Delphinidin glucoside	Anthocyanin	PE	8.128	803.2019	Positive	Antioxidant	Watsona & Schönlau 2015
45	1-Eicosene	Hydrocarbon	PP/PE	8.240	280.3130	Positive	Antibacterial	Lay-Jing et al. 2012
46	Spermine derivates	Polyamine	PP/PE	8.302	370.2721	Positive	Tyrosinase inhibitor	Kim et al. 2018
47	Scytonemin	Alkoid	PE	8.474	544.1423	Positive	Antioxidant and tyrosinase inhibitor	He et al. 2022
48	Diisobutyl/diisononyl phthalate	Benzene carboxylic acid	PP/PE	8.537	278.1521	Positive	Antibacterial	Huang et al. 2021
49	10-Gingerol	Phenol	PE	8.586	350.2457	Positive	Antioxidant	Lee & Ahn 1985
50	Tianeptine		PE	8.633	436.1233	Negative	Antioxidant	Dello et al. 2012
51	Luteolin derivate	Flavonoid	PP/PE	8.686	710.1720	Negative	Tyrosinase inhibitor and Antibacterial	Zhang et al. 2017
52	Nonoxynol-9	Alkane	PP/PE	8.817	616.4182	Positive	Antibacterial	Hillier et al. 2005
53	Annosquamosin B	Diterpenoid	PE	9.268	308.2351	Positive	Antioxidant	Pandey & Barve 2011
54	Aeglin	Fatty acyl glycoside	PE	9.324	510.2095	Positive	Antioxidant	Patel et al. 2020
55	Rubroskyrin	Bis-anthroquinone	PP	9.373	574.1103	Negative	Antioxidant and tyrosinase inhibitor	Bräse et al. 2012
56	Quercetin-3- glucoside	Flavonoid	PP	9.466	714.1554	Positive	Antioxidant and tyrosinase inhibitor	Chen & Kubo, 2002, Heo et al. 2007, Baghel et al. 2012
57	Harderoporphyrin	Heterocyclic m crocycle	PP/PE	9.761	608.2627	Positive	Tyrosinase inhibitor	Baldea et al. 2016
58	Pheophorbide A	Heterocyclic compound	PE	10.085	592.2687	Positive	Antioxidant and tyrosinase inhibitor	Lanfer et al. 2005, Li et al. 2012b
59	Ginsenoside Rh4	Saponin	PE	10.152	620.4282	Positive	Tyrosinase inhibitor	Kim 2015
60	Melilotoside A	Hydroxycinnamic acid derivate	PE	10.722	590.4177	Positive	Antioxidant and tyrosinase inhibitor	Yang et al. 2007; Güven et al. 2023
61	Paradol	Phenol	PE	10.890	334.2504	Positive	Antioxidant and tyrosinase inhibitor	Nile & Park 2015, Viggiano et al. 2023
62	Biotin	Heterocyclic compound	PE	11.259	614.4155	Positive	Antioxidant	Al-Quadah & Ismail 2012
63	Methenamine	Heterocyclic compound	PE	12.791	140.1063	Positive	Antioxidant	Wetchakul et al. 2022

Enzyme inhibitory activities in lichens have been studied before, and interesting results were obtained. Of these, lichens showing tyrosinase inhibitory activity were *Flavoparmelia caperata*, *Letharia vulpina*, *Cetraria juniperina*, *Parmotrema perlatum* (Malaspina et al. 2020), *Hypogymnia physodes*, *Bulbothrix setschwanensis*, *Usnea* species (Zhao et al., 2021), *Cetraria islandica* (Akbulut & Yildiz 2010) and *Himantormia* (Areche et al. 2022) species. However, studies of tyrosinase inhibitory activity in *Peltigera* lichens have not been done before. Since many substances showing tyrosinase inhibitory activity were detected in our study, the activities of the extracts were examined and determined as 86.95 and 196.7 µg/mL (IC_{50}) in PP and PE, respectively. Standard compound kojic acid's inhibitory activity was 37,40 µg/mL

(IC_{50}) (Table IV). Compared to kojic acid, PP and PE's IC_{50} values were found to be close and that indicated the strong tyrosinase inhibitor activity.

According to antibacterial and antifungal compounds C16 sphinganine, tetradecyl amine, diisobutyl phthalate, nonoxynol-9, erythronolide B, 14, 19- dihydroaspidospermatine, fenpiclonil, eudesmin, 1- eicosene, 9- nonadecene, triclofoteolin and 8-hydroxyluciferol extracts, which are known to exist in both lichens, extracts antibacterial and antifungal effects were tested. They did not show activity against any of the strains. It was observed that similar results were obtained in antibacterial tests performed on *Peltigera* species before. It is thought that this may be due to the synergistic effects of different substances in the extract.

Table II. TPC, TFC and *in vitro* antioxidant activity results.

	(mg GAE/ g extract) ^a	(mg QE/ g extract) ^b	(IC_{50}) (µg/ mL) ^c	(IC_{50}) (µg/ mL) ^d	(IC_{50}) (µg/ mL) ^e	(IC_{50}) (µg/ mL) ^f	(TEAC _{CUPRAC}) ^g
PP	29,81±4,3	32,87±1,8	84,55±0,34	51,27±0,10	95,01±0,23	20,57±0,19	4,69±0,11
PE	168,47±4,1	34,88±1,5	9,349±0,14	9,127±0,08	58,65±0,26	70,08±0,11	9,57±1,1
α-tocopherol							
Trolox				18.87±0.01			
Ascorbic Acid			4.474 ± 0.31		5.992± 0.01		
EDTA						6,619±0,21	

^aResults of TPC. ^bResults of TFC. ^cResults of DPPH assay. ^dResults of ABTS assay. ^eResults of Superoxide scavenging activity. ^fResults of Metal chelating assay. ^gResults of CUPRAC assay.

Table III. CUPRAC slope values.

	slope
Trolox	0,052
PP	0,2439
PE	0,4975

Table IV. Tyrosinase inhibitor activity results.

	Tyrosinase inhibitory activity(IC_{50}) (µg/mL)
PP	86,95±0,11
PE	196,7±0,22
Kojic acid	37.40 ± 0.42

Table V. Statistical Analysis (One way ANOVA).

Activity	Sum of Squares	df	Mean Square	F	Sig.
Between Groups	10344.257	3	3448.086	11966.634	.000
Within Groups	2.305	8	.288		
Total	10346.562	11			

CONCLUSIONS

It has been determined that PP and PE lichens contain lichen phenolics, alkaloids, terpenic metabolites and flavonoids. They showed antioxidant and tyrosinase inhibitor activities, which are among the expected effects in line with the substances they contain. Although many substances with antibacterial and antifungal properties were detected, they were found to have no effect, suggesting that this result was due to the synergistic effect of the substances contained in it. It was determined that PP and PE have strong antioxidant and tyrosinase inhibitor activities and it was envisaged that further studies could be conducted. *Peltigera* lichens showed antioxidant and tyrosinase inhibitory activity in line with the secondary metabolites but showed no antimicrobial effect.

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SUPPLEMENTARY MATERIAL

Table S1.

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Author contribution

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